






A. Definitions





1. Explosion





1. Explosion

The sudden conversion of potential energy (chemical or mechanical) into kinetic energy with the production and release of gases under pressure, or the release of gas under pressure. These high pressure gases then do mechanical work such as moving, changing, or shattering nearby materials. (NFPA-921)





Elements of an Explosion

- Rapid Increase in Gas Pressure (Gas Dynamic)
- Confinement of the Pressure
- Rapid release of that Pressure
- Damage or Change to the confining structure or the vessel
- Noise is not an element (Not required)



2. Explosives

The term “explosives”, generally is used in reference to a wide range of energetic materials that can react chemically to produce heat, light, and gas.



3. Combustion Explosion

The rapid combustion of a fuel in a confined area.

- Fuel Gases
- Industrial Gases
- Dust





4. Deflagration

Rapid burning.

Deflagration is a rapid chemical reaction in which the output of heat is enough to enable the reaction to proceed and (depending on the ambient conditions of the fuel) be accelerated without input of heat from another source. (Subsonic Reaction)



5. Detonation

Instantaneous combustion or conversion of a solid, liquid or gas into larger quantities of expanding gases accompanied by heat, shock and most often a noise. (Supersonic Reaction)



6. Detonation Velocity

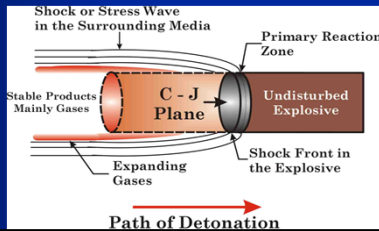




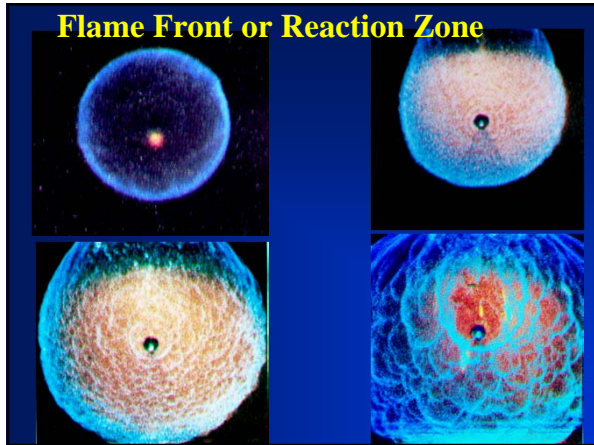
a. Detonation Wave

A detonation wave is a shock wave in a reacting (explosive or fuel) material where the chemical reaction is carried out in the shock front.

Example is an explosive material. However, in a diffuse fuel the flame front or reaction zone is also present.



Flame Front or Reaction Zone



7. Deflagration to Detonation Transition (DDT)

Once a self-sustaining reaction has begun, it propagates through the adjacent material at a rate determined by either porosity, particle size, density, pressure, heat, and distance.

Deflagration(Subsonic) transitions to Detonation (Supersonic) reaction rate.



B. Types Of Explosions



1. Mechanical Explosion

Nature of the fuel
does not change.



1. Mechanical Explosion





2. Chemical Explosion

Nature of the fuel changes.

Explosives, no oxidizer required



2. Chemical Explosion

Nature of the fuel changes

Combustion, Oxidizer Required



3. Electrical Explosion

An electrical explosion is caused by a high-energy electrical arc which generates sufficient heat to cause failure of the containing component.





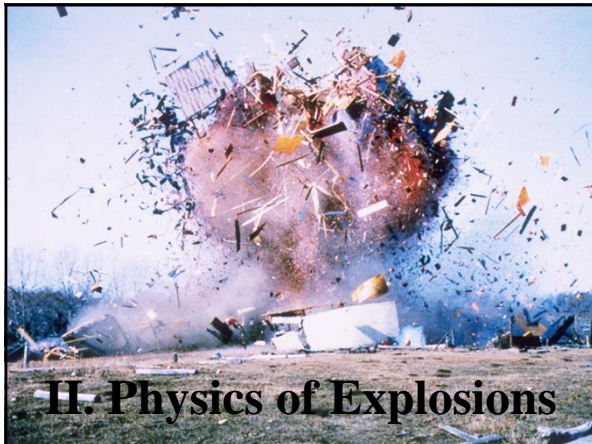
3. Electrical Explosion



4. Nuclear Explosion

An atomic explosion is induced by
either fission or fusion.





II. Physics of Explosions



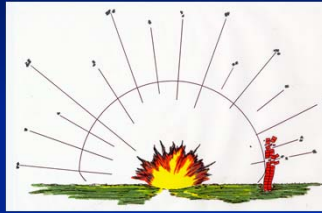
A. Effects of an Explosion

Blast Pressure

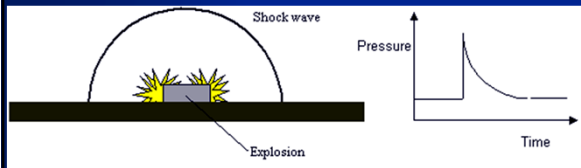
Fragmentation

Thermal or Incendiary

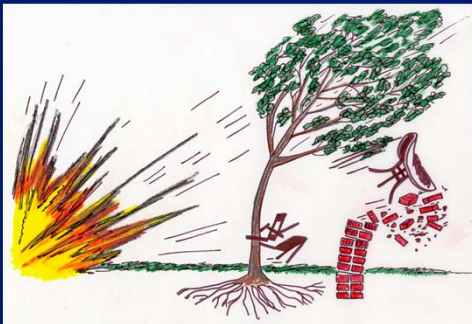
Ancillary Effects



1. Blast Pressure Effects

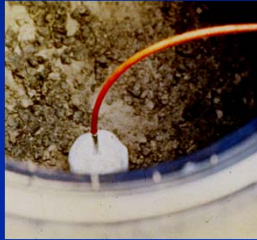


a. Positive Pressure Phase





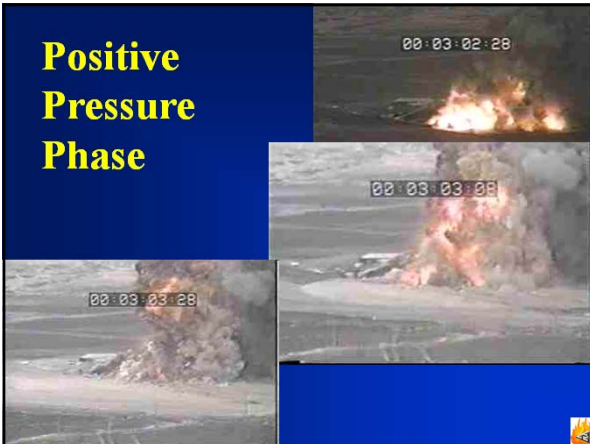
Positive Pressure Phase



Positive Pressure Phase



Positive Pressure Phase





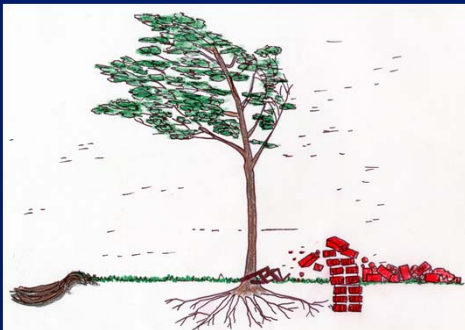
Positive Pressure Phase

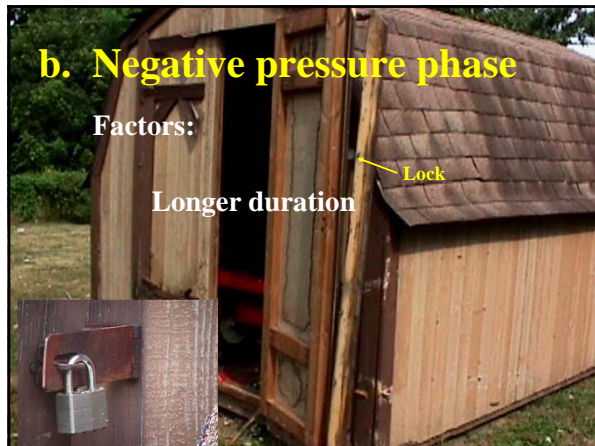
PEPCON Rocket Fuel
Fire and Explosion, 1988

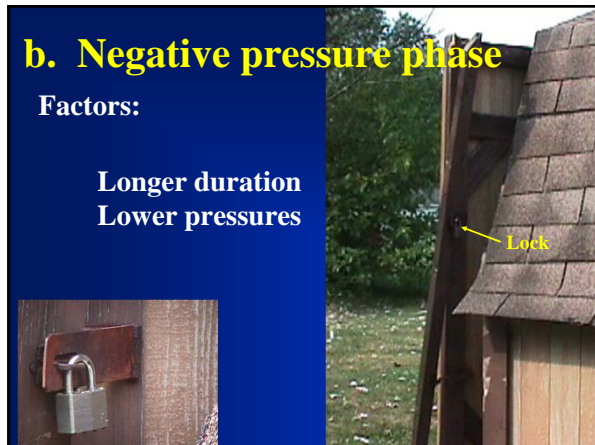


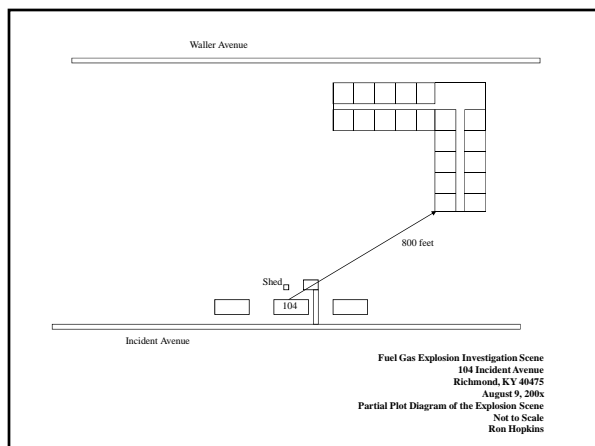


b. Negative pressure phase











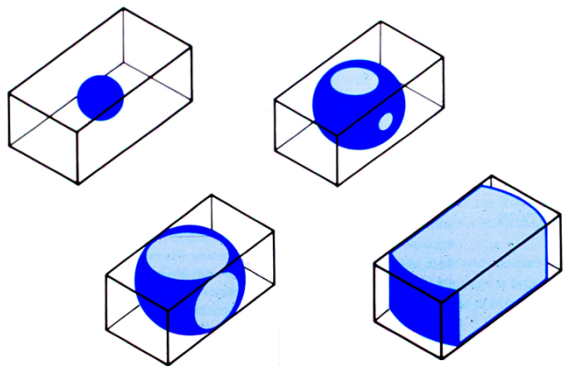
b. Negative pressure phase

Factors:

- Longer duration
- Lower pressures
- Lower air velocity



c. Shape of the Pressure Wave

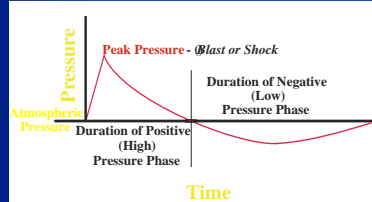




d. Rate of Pressure Rise

v.

Maximum Pressure



2. Fragmentation Effect

Fragments

Shrapnel



2. Fragmentation Effect

Fragments

Shrapnel





3. Thermal Effect - Incendiary

Example temperatures: 3000 – 7000 degrees F.

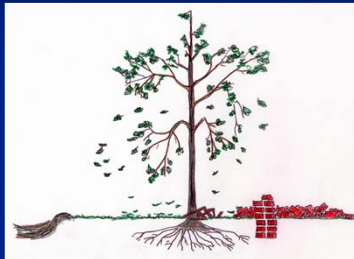


High Density and Low Density Fuels



4. Secondary Blast Pressure Or Ancillary Effects

a. Reflection





4. Secondary Blast Pressure Or Ancillary Effects



a. Reflection

b. Earth, Water Shock, and Ceiling

D. Factors Controlling Explosion Effects



1. Nature of the Fuel and Oxidizer
2. Quantity of the Fuel Present
3. Configuration of the Fuel
4. Blast Pressure Front Modifiers
4. Containment Vessel
5. Initiation Source and Location
6. Venting

F. Seated Explosion

The “seat” of an explosion is defined as the crater or area of greatest damage located at the point of initiation (epicenter) of an explosion.





a. Condensed Phase Fuel

An explosive material in the form of a solid or liquid rather than a gas or vapor.



b. Mechanical Explosion

Boiling
Liquid
Expanding
Vapor
Explosion





G. Non-Seated Explosion

Those explosions where there is no physical evidence of a single location where the explosion originated.



Diffuse Phase Fuel

a. Diffuse Phase Fuel, Gases and Vapors

A general category of combustion explosions that occur as a result of the ignition of fuel gases (i.e. Natural Gas, LPG), Industrial Gases, Sewer Gases, and vapors of pooled liquids (i.e. gasoline vapors, lacquer thinner, MEK).



b. Dust Explosion

Ignition of solid materials such as dusts and fines.





III. Explosives

A. General

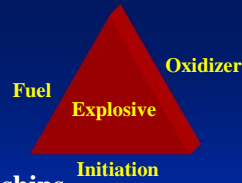
Fuel Oxidizer Relationships

Explosive

Fuel

Oxidizer - Oxygen Balance

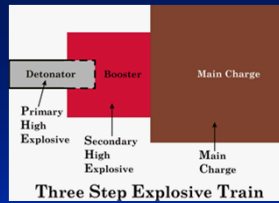
Special Fuel or Sensitizer



1. Definitions

a. Firing Train

A sequence of events required to initiate a single or final event.



b. High Order Detonation

Complete burning of the explosive material or initiation of the material at maximum velocity.





c. Low Order Detonation

Incomplete burning of the explosive material or initiation of the material at less than maximum velocity.

Reaction at less than maximum velocity



2. Low Explosives

- a. Deflagrates
- b. Material is a mixture
- c. Initiated by heat.
- d. Confinement required for explosion
- e. VOE below 3000 ft/sec

Examples: Black powder, smokeless powder



3. High Explosives

- a. Material undergoes detonation without confinement.
- b. Material is a compound
- c. Initiated by shock or heat
- d. Supersonic reaction in the product.
- e. High brisance
- f. VOD above 3300 ft/sec

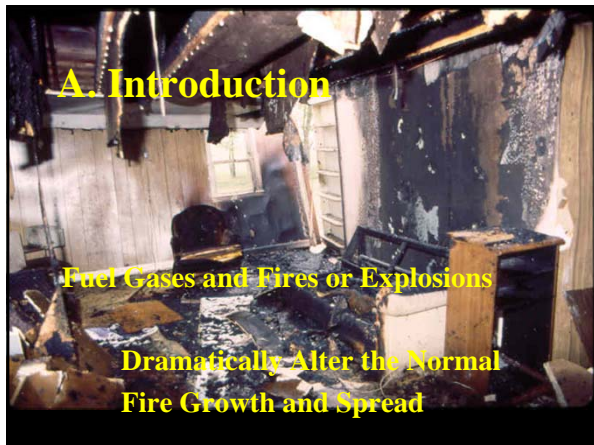
Examples: Dynamite, ANFO, PETN





IV. Diffuse Fuel Explosions

- Fuel Gases
- Pooled Flammable/Combustible Liquids
- Dusts
- Backdraft



A. Introduction

Fuel Gases and Fires or Explosions
Dramatically Alter the Normal
Fire Growth and Spread

NFPA 54 National Fuel Gas Code

From the “point of delivery” to the
connections with each gas utilization
device

The “point of delivery” shall be considered the outlet of
the service meter assembly or the outlet of the service
regulator or service shutoff valve where no meter is
provided.



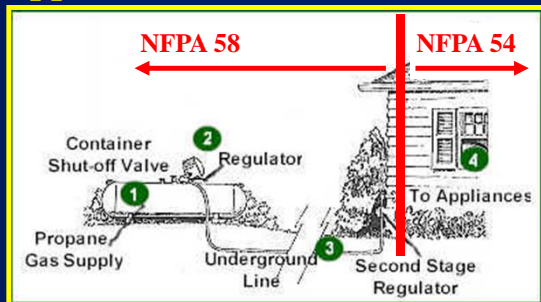
NFPA 58

Liquefied Petroleum Gas Code

Containers, piping, and associated equipment, when delivering LP-Gas to a building for use as a fuel gas.

Including tanks, cylinders, and piping up to the second stage regulator

Application of the NFPA Codes



B. Fuel Gases

Fuel gases by definition:

- Natural Gas (Commercial)
- Liquefied Petroleum Gas (in the vapor phase only)
- Liquefied Petroleum Gas–Air mixtures
- Manufactured Gases
- Mixtures of these gases

Most commonly encountered by the fire and explosion investigator will be natural gas and commercial propane.





1. Odorization

- LP-Gas and natural gas have little or no identifiable odor in their natural state
- Odorant containing t-butyl mercaptan, thiophane, ethyl mercaptan or other mercaptans are added by law
 - Natural Gas - 49 CFR 129.625
 - LP gas NFPA 58 Section 4.2.1
- Must be noticeable **“at concentration in air of one-fifth of the lower explosive limit”**

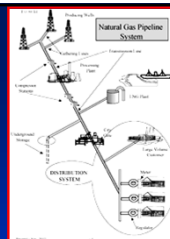
1. Odorization

- **Natural gas odorant** is added by the local distribution company prior to the introduction of the gas into the distribution Pipelines (mains).
 - Natural gas in long-distance transmission pipelines is usually not odorized.
- **LP-Gas odorant** is added by the gas supplier prior to delivery to an LP-Gas distributor's bulk plant.

2. Fuel Gas System Components

a. Natural Gas Systems

Typically piped directly to the consumers' buildings from centralized production and storage facilities via:
Transmission Pipelines.
Distribution Pipelines (Mains)





b. Fuel Gas System Components

Natural Gas Systems

Service Lines (House Lines)

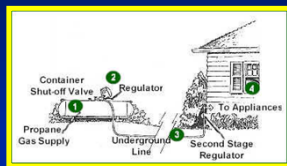
- Piping
- Pressure regulation
- Metering
- Valving
- Utilization equipment



Fuel Gas System Components

b. LPG Systems

- Storage Tank or Cylinder
- Piping
- Pressure regulation
- Metering
- Valving
- Utilization equipment



(1.) LP-Gas Storage Containers

ASME Tanks (>120 Gallons)

ASME Boiler and Pressure
Vessel Code



DOT Cylinders (<120 Gallons)

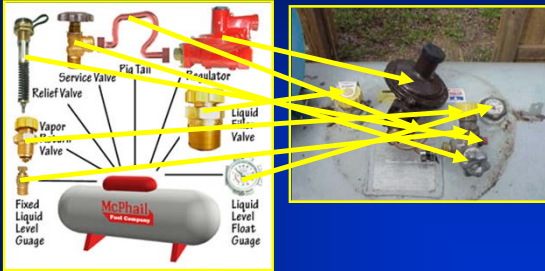
49 CFR - Transportation





(2.) Container Appurtenances

(Tanks)



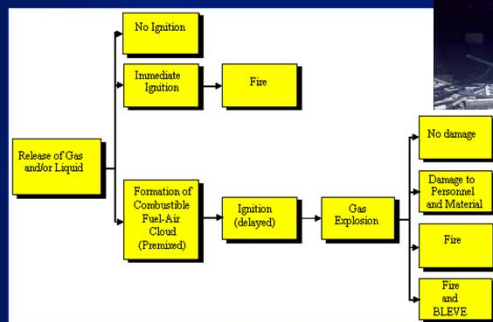
c. Normal Working Pressures

Natural Gas 8" W.C. (~ 0.3 PSI)
Propane 11" W.C. (~0.4 PSI)

Some appliances have additional
regulators to lower working pressures
to about 3.5" W.C. (~0.13 PSI)

1 psi = 2.7" W.C.

3. Fuel Gas Explosions





Gas Migration Ignition Study, Minnesota Chapter IAAI



To much gasoline!

Flash fire or unconfined combustion
explosion?



FUGLY.COM



4. Characterization of Explosion Damage



a. High Order Damage



Characteristics

- Small Debris Pieces
- Long Missile Distances
- Fast Rate of Pressure Rise
- Negative Pressure Phase is powerful





Fuel - Air Relationship



LEL UEL

- Optimum Mixture
- Near or just above stoichiometric
- Most efficient burning
- Little following fire

b. Low Order Damage



Characteristics

- Large Debris
Pieces



- Short Missile
Distances



- Slower Rate of
Pressure Rise

- Pushing or Heaving



Fuel - Air Relationship

- Near LEL or UEL
- Inefficient burning
- Low rate of pressure rise
- Low speed pressure wave
- Near LEL, little following fire
- Near UEL, greater potential for following fire

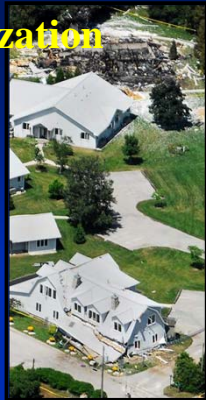


LEL

UEL



Damage Characterization







c. Vapor Density and Damage



Relationship of Gas in Compartment?
Why?

Lighter-than-air gases

- Collect in upper areas
- Pocketing at ceilings
- Migrate through openings



Heavier-than-air gases

- Collect in lower areas
- Burns at high levels when ignited
- Low pocketing is unusual





**c. Location of
damage is not
indicative of vapor
density**



A common misconception

- More a function of wall strength or,
- Height of explosive range





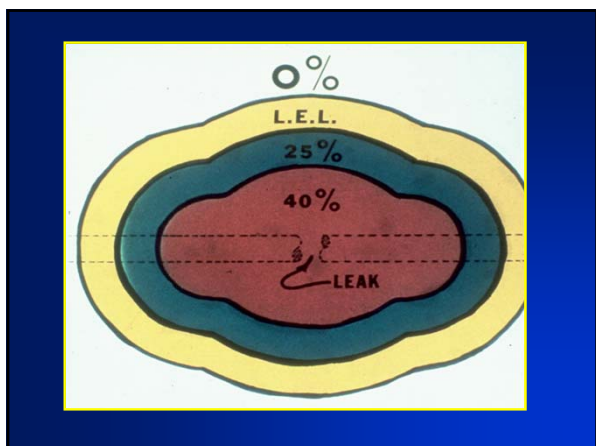


d. Minimum Ignition Energy

- Most easily ignitable fuels
- Ignition Temperatures
700 - 1100 F
- Ignition Energies 0.20 - 0.25 millijoules
Examples would include:
Static Electricity,
Operation of Motors, Switch

e. Underground Migration *Fuel Gases*







Gas Line Bar Hole Survey

f. Multiple ("Cascade") Explosions

- a. Multiple pockets of gas
- b. "Cascade" from room to room or floor to floor
- c. Aeration of pockets over the UEL
- d. Multiple explosions are very common





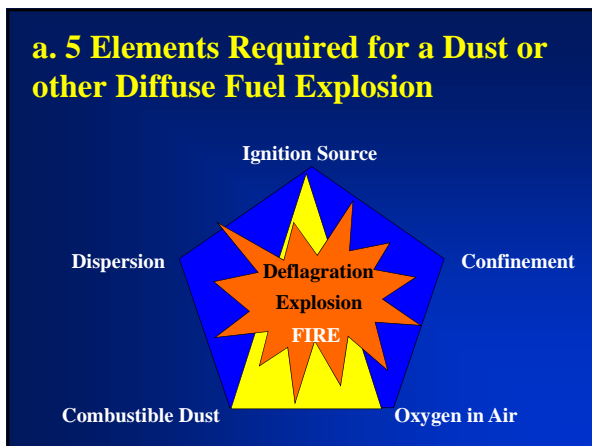
E. Dust Explosions



1. Dynamics of Dust and other Diffuse Fuel Explosions



a. 5 Elements Required for a Dust or other Diffuse Fuel Explosion





b. Diffuse Phase Fuels, Dust



b. Wide variety of materials

Combustible and Non-Combustible



c. Controlling Factors

- Suspended or Layered
- Particle Size
- Concentration
- Turbulence
- Moisture





2. Progression of Dust Explosions

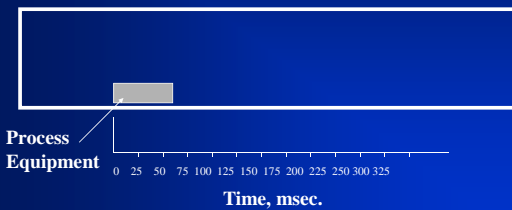
- a. Usually occur in series
- b. Initial explosions usually less violent than subsequent
- c. Subsequent explosions are fueled by additional dust put into suspension



3. Typical Dust Explosion Event

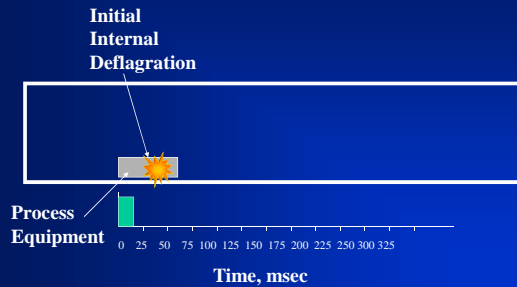


Progression of a Dust Explosion

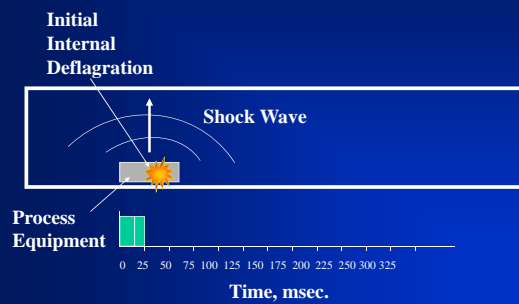




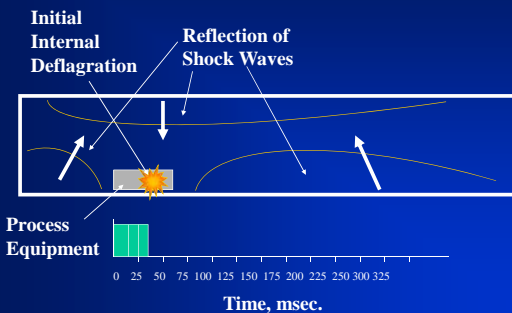
Progression of a Dust Explosion



Progression of a Dust Explosion

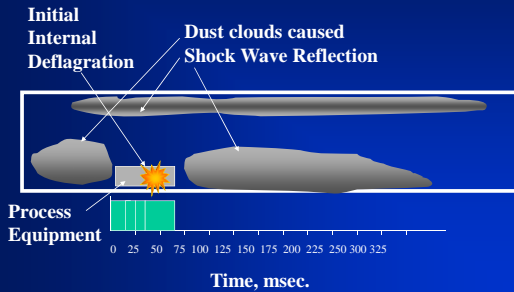


Progression of a Dust Explosion

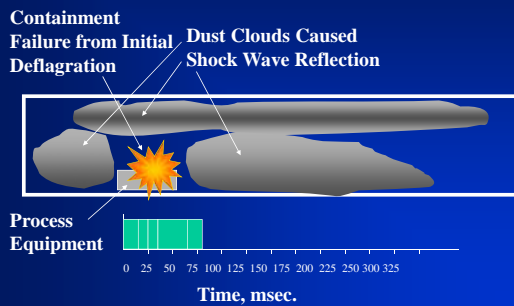




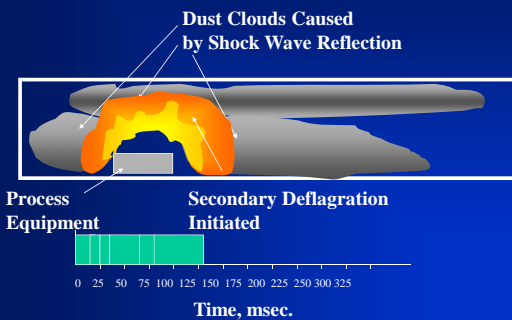
Progression of a Dust Explosion

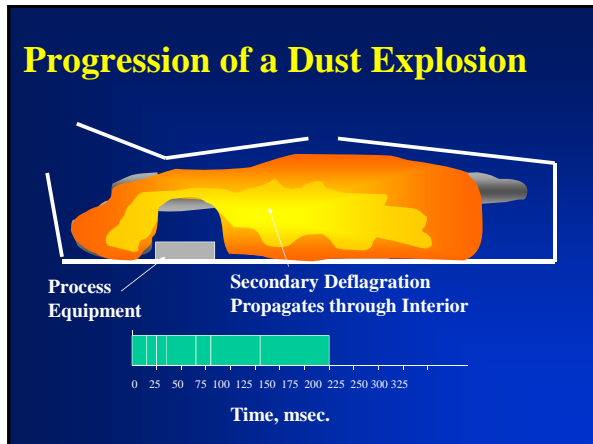


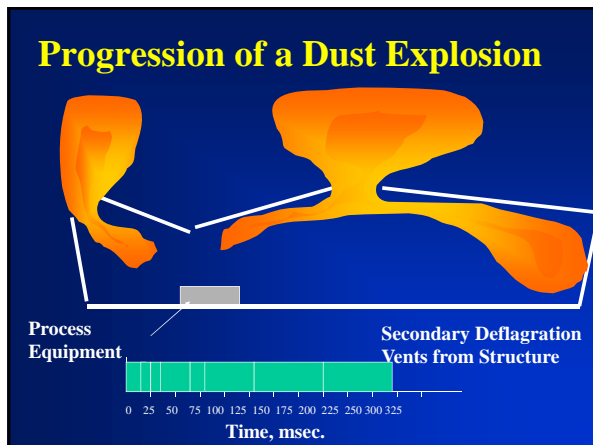
Progression of a Dust Explosion

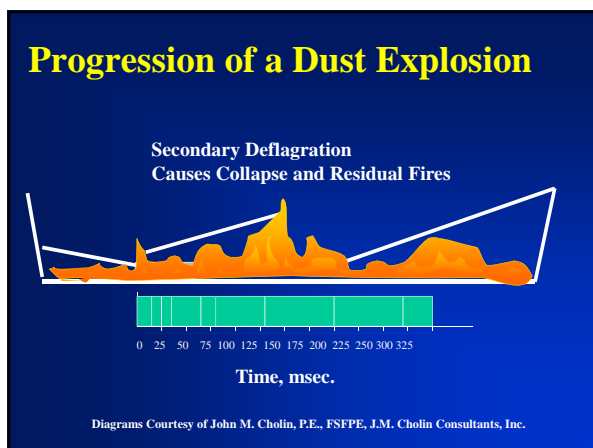


Progression of a Dust Explosion











CSB Model of the Imperial Sugar Plant Explosion



G. Backdraft Explosions

1. Ventilation Controlled Fires *Post Flashover Conditions*













B. Systematic Approach is Even More Important



1. Scenes are usually larger than simpler fires
2. Scenes are usually more disturbed than fires

C. Secure the Scene

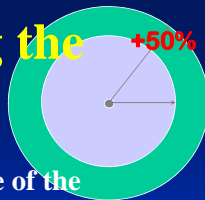
Establish and maintain control of the scene and area



Prevent unauthorized persons from entering



D. Establishing the Scene



1. 1 1/2 times the distance of the furthest piece of debris
2. Debris may have been propelled great distances into adjacent buildings or vehicles
3. As additional debris is found, the scene is widened



E. Scene Search

1. Outer perimeter inward towards epicenter



2. Briefing and Control of Search Teams



Identifying evidence

Photographing evidence

Mapping evidence

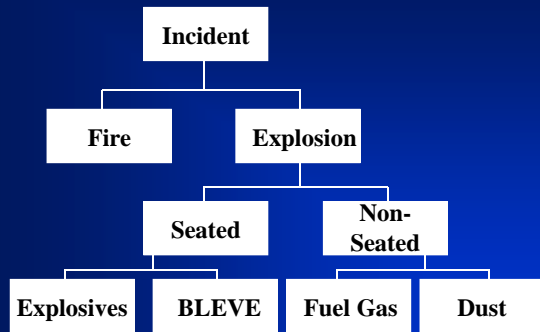
3. Safety at the Explosion Scene



- a. Structures are unsound
- b. Secondary explosions are possible
- c. In bombings, secondary devices, unexploded devices or undetonated explosives are possible
- d. Special Scene Hazards



F. Incident Assessment

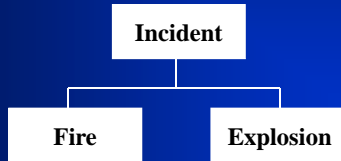


1. Initial Incident Assessment

Identify Explosion or Fire

Burning or Heat Treatment

Overpressure







a. Low or High Order Damage



b. Seated or Non-Seated



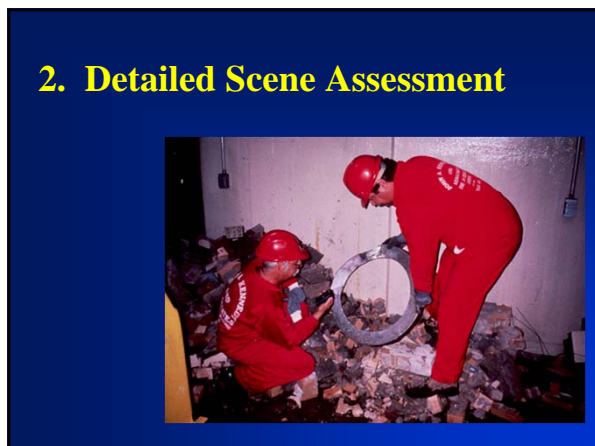
c. Type of Explosion

- Mechanical
- Chemical
- Combustion
- BLEVE
- Electrical











**a. Record the Investigation,
Evidence and
Findings**



- Notetaking
- Diagramming and Mapping
- Photography
- Proper Evidence Collection and Preservation

**b. Identify Damage Effects
of the Explosion**

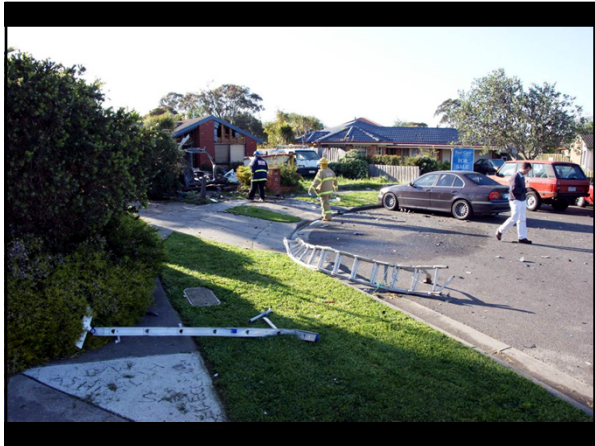


Note each instance of blast or overpressure damage

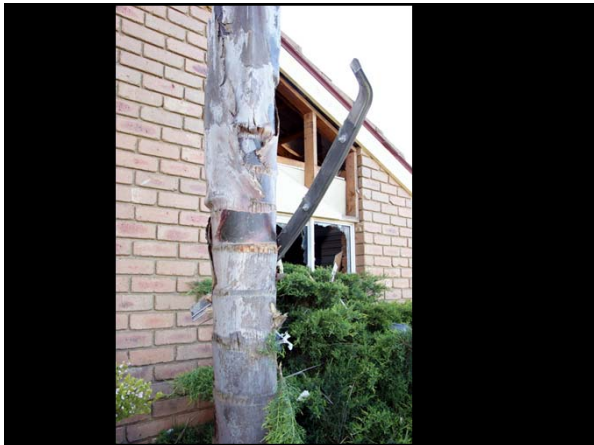


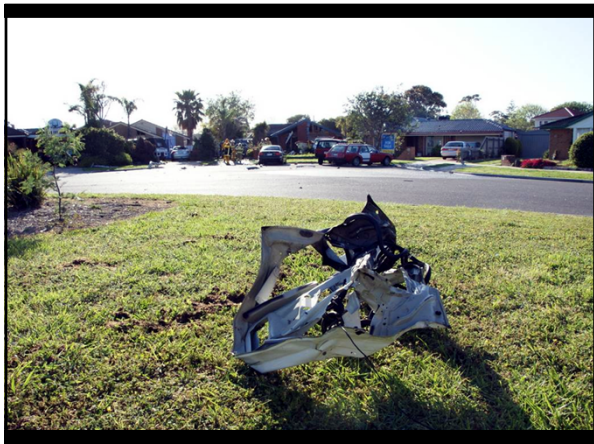




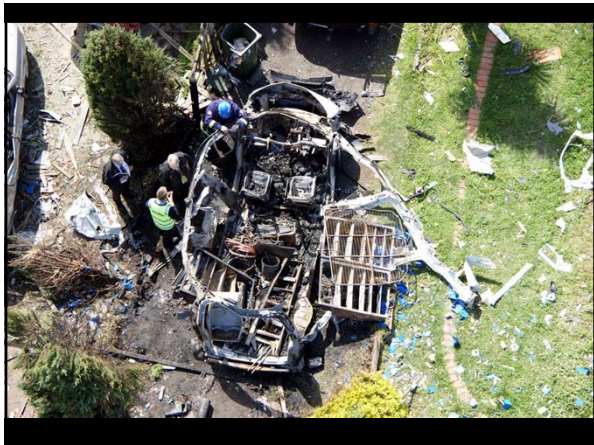


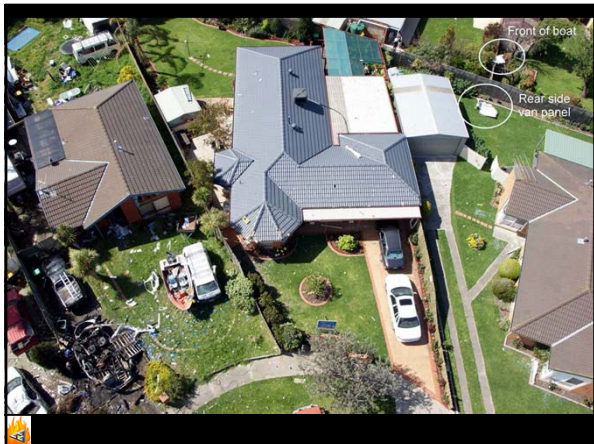














c. Identify Pre-Blast and Post-Blast Fire Damage

Propelled Debris may be burned or unburned



d. Locate, Identify and Record Articles of Evidence



Evidence may have been propelled into a variety of locations



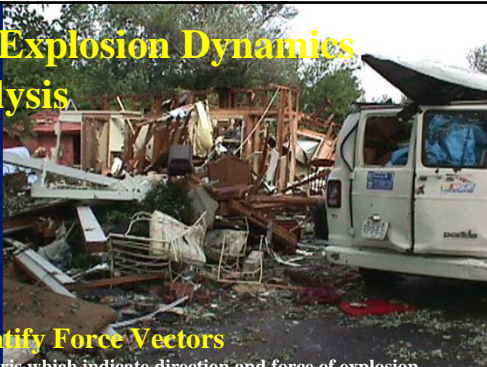


VI. Explosion Dynamics Analysis

A. Identify Force Vectors

Note debris which indicate direction and force of explosion

Direction
Magnitude



B. Identify epicenter

Exact epicenter most often found with seated explosions

Non-seated explosions produce larger origins



C. Analyze Origin (Epicenter)

1. Explosion Dynamics Analysis

Trace Force Vectors

Least to Most Damage





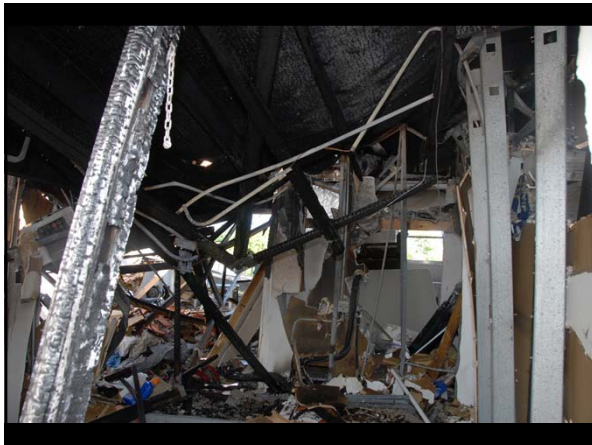
2. Construct Explosion Dynamics

Vector Diagram

Direction of debris movement

Relative force of debris movement

Both large scale and small scale
diagrams may be necessary









D. Analyze Fuel Source

1. Compare nature of damage to available fuels

2. All available fuels must be considered and eliminated

3. Samples





4. Physical Evidence Samples

a. Residues

Ignitable Liquids

Explosives

b. Fuel Containers/ Appliances/ Equipment

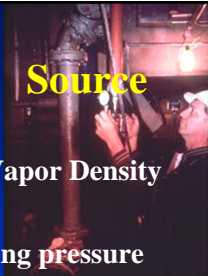


5. Determine Fuel Source

Source may be related to Vapor Density

All gas appliances and piping pressure tested

Any leaks discovered must be identified as pre- or post-blast



E. Analyze Ignition Source

1. Often most difficult

2. Multiple possible ignition sources often present





3. Consider all available information

Minimum Ignition Energy of Fuel
Ignition Energy of Ignition Source
Ignition Temperature of Fuel
Temperature of Ignition Source
Location of Ignition Source in Relation to Fuel
Contemporaneous presence of Fuel and Ignition Source
Witness Accounts

F. Analyze Cause/Responsibility

1. What brought together Fuel and Ignition Source at the Origin

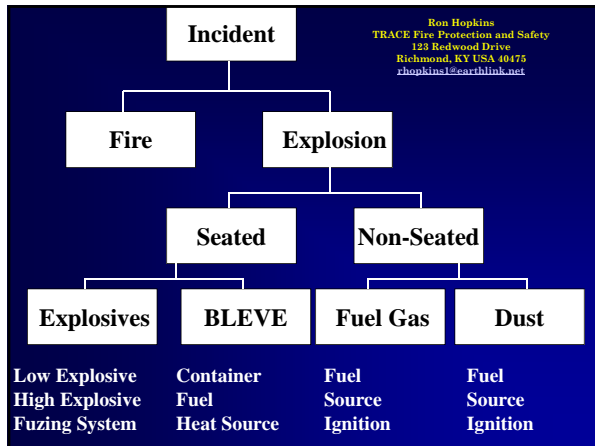
Action
Omission
Circumstances

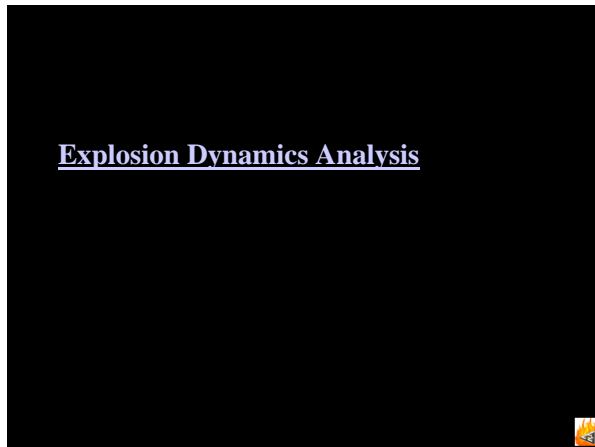


2. What could have prevented the Explosion

- Compliance to Codes
- Compliance to Standards
- Compliance to Good Practice
- Proper Industrial Engineering












Questions?

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